



# RESEARCH & ANALYTICAL LABORATORIES, INC.

Analytical/Process Consultations

07 April 2011

Compliant Service Branch  
Public Water Supply Section  
1634 Mail Service Center  
Raleigh, NC 27699-1634  
Attn: Christyn Fertenbaugh

Re: Corrosion Control Evaluation  
PWSID No NC0241410  
Brown Summit Middle School  
Guilford County Schools

Dear Ms. Fertenbaugh:

Enclosed please find the Corrosion Control Treatment Study (CCT) for the above referenced facility. This groundwater system has exceeded the copper action level (i.e. 2.97 mg/l –October 2007). It is important to note that this water supply system has been in lead and copper compliance for the last four (4) years and should be eligible for reduced monitoring. It is suspected that inadequate flushing may have been the cause of the 2007 copper level exceedance. It is therefore recommended that the facility apply for reduced monitoring, carefully monitored (i.e. pH, copper, etc.), and no further action be taken until the lead or copper exceedance can be documented. These recommendations are based upon current and historical lead and copper data, water quality parameters, and the US EPA Revised Guidance Manual for Selecting Lead and Copper Control Strategies.

If you should have any questions concerning this report, or need additional information please so advise.

Sincerely,

James M. Cheshire  
President/CEO  
Research & Analytical Labs, Inc.

JMC/js

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# Evaluation Form for Corrosion Control Treatment

MAIL TO: Compliance Branch  
Public Water Supply Section  
1634 Mail Service Center  
Raleigh, North Carolina 27699-1634

## A. PWS General Information

Date: 04/05/2011

1. PWS Name: Brown Summit Middle School
2. PWSID Number: NC0241410
3. Contact Person:  
Name: Mr. Larry Odell  
Mailing Address: Brown Summit Middle School  
3920 Naco Road  
Greensboro, North Carolina 27405  
Telephone: 336-370-2387  
Fax: 336-370-2398
4. Population served: 236
5. Person Responsible for preparing this form:

Name: James M. Cheshire  
Telephone: 336-996-2841

Signature: 

## PWS Technical Information

1. Monitoring Results:  
Sampling Dates: From 07/01/07 To 12/31/07

### First-Flush Tap Monitoring Results:

Lead:  
Minimum concentration = <0.003 mg/L  
Maximum concentration = <0.003 mg/L  
90th percentile = <0.003 mg/L

Copper:  
Minimum concentration = 1.24 mg/L  
Maximum concentration = 4.05 mg/L  
90th percentile = 2.97 mg/L

Point of Entry Monitoring Results (Averages): Water Quality Parameters (March 28 & 29 2011)

	Points of Entry				
	1	2	3	4	5
Lead Concentration in mg/L:	<u>&lt;RRL</u>	<u>      </u>	<u>      </u>	<u>      </u>	<u>      </u>
Copper Concentration in mg/L:	<u>&lt;RRL</u>	<u>      </u>	<u>      </u>	<u>      </u>	<u>      </u>
pH:	<u>6.3</u>	<u>      </u>	<u>      </u>	<u>      </u>	<u>      </u>
Temperature, °C:	<u>12</u>	<u>      </u>	<u>      </u>	<u>      </u>	<u>      </u>
Alkalinity, mg/L as CaCO <sub>3</sub> :	<u>20</u>	<u>      </u>	<u>      </u>	<u>      </u>	<u>      </u>
Calcium, mg/L as Ca:	<u>2.84</u>	<u>      </u>	<u>      </u>	<u>      </u>	<u>      </u>
Conductivity, $\mu$ mho/cm @ 25° C:	<u>83</u>	<u>      </u>	<u>      </u>	<u>      </u>	<u>      </u>
Phosphate, mg/L as P:	<u>&lt;RRL</u>	<u>      </u>	<u>      </u>	<u>      </u>	<u>      </u>
Silicate, mg/L as SiO <sub>2</sub> :	<u>7.53</u>	<u>      </u>	<u>      </u>	<u>      </u>	<u>      </u>

\*DIC = 9 mg C/L (saturation pH = 9.5)

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Water Quality Parameter Distribution System Monitoring Results:  
Indicate whether field or laboratory measurement.

	Field	Lab
pH: minimum = <u>6.3</u> maximum = <u>6.5</u>	<u>X</u>	_____
Alkalinity: minimum = <u>19</u> mg/L as CaCO <sub>3</sub> maximum = <u>20</u> mg/L as CaCO <sub>3</sub>	_____	<u>X</u>
Temperature: minimum = <u>13</u> °C maximum = <u>18</u> °C	<u>X</u>	_____
Calcium: minimum = <u>2.80</u> mg/L as Ca maximum = <u>2.87</u> mg/L as Ca	_____	<u>X</u>
Conductivity: minimum = <u>83</u> µmho/cm @ 25° C maximum = <u>84</u> µmho/cm @ 25° C	_____	<u>X</u>
Orthophosphate: (if phosphate based inhibitor is used) minimum = <u>&lt;RRL</u> mg/L as P maximum = <u>&lt;RRL</u> mg/L as P	_____	<u>X</u>
Silica: (if silica based inhibitor is used) minimum = <u>7.49</u> mg/L as SiO <sub>2</sub> maximum = <u>7.52</u> mg/L as SiO <sub>2</sub>	_____	<u>X</u>

2. Existing Conditions:

Is treatment used? ☒ yes ☐ no

Identify water source(s):

Source No. 1 P01 (Well)

Source No. 2 \_\_\_\_\_

Source No. 3 \_\_\_\_\_

If treatment is used, is more than one source used at a time?

☐ yes ☒ no

Identify treatment processes used for each source:

<u>Process</u>	<u>No. 1</u>	<u>No. 2</u>	<u>No. 3</u>
Presedimentation	_____	_____	_____
Aeration	_____	_____	_____
Chemical mixing	_____	_____	_____
Flocculation	_____	_____	_____
Sedimentation	_____	_____	_____
Recarbonation	_____	_____	_____

Identify treatment processes used for each source:

<u>Process</u>	<u>No. 1</u>	<u>No. 2</u>	<u>No. 3</u>
2nd Stage mixing	_____	_____	_____
2nd Stage flocculation	_____	_____	_____
2nd Stage sedimentation	_____	_____	_____
Filtration:	_____	_____	_____
Single medium	_____	_____	_____
Dual media	_____	_____	_____
Multi-media	_____	_____	_____
GAC cap on filters	_____	_____	_____
Disinfection:	_____	_____	_____
Chlorine	<u>X</u>	_____	_____
Chlorine dioxide	_____	_____	_____
Chloramines	_____	_____	_____
Ozone	_____	_____	_____
Granular Activated Carbon	_____	_____	_____

List chemicals normally fed:

List chemicals sometimes fed:

## 3. Present Corrosion Control Treatment:

None X

Inhibitor \_\_\_\_\_

Date initiated \_\_\_\_\_

Present dose \_\_\_\_\_ mg/L

Range in Residual in Distribution System:

Maximum \_\_\_\_\_ mg/L Minimum \_\_\_\_\_ mg/L

Brand name \_\_\_\_\_

Type \_\_\_\_\_

Has it been effective? Please comment on your experience.

pH/alkalinity adjustment \_\_\_\_\_

pH Target \_\_\_\_\_

Alkalinity Target \_\_\_\_\_ mg/L CaCO<sub>3</sub>

Calcium adjustment \_\_\_\_\_

Calcium Target \_\_\_\_\_ mg/L CaCO<sub>3</sub>

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## 4. Water Quality:

Complete the table below for typical untreated and treated water quality data. Copy this form as necessary for additional sources. Include data for each raw water source, if surface supplies are used, and finished water quality information (point of entry) from each treatment plant. If wells are used, water quality information from each well is acceptable but not necessary if several wells have similar data. For groundwater supplies, include a water quality summary for each wellfield or grouping of wells with similar quality.

Include available data for the following:

Parameters	Untreated Supply	Treated Water (point of entry)
pH, units	N/A	6.29, 6.35
Alkalinity, mg/L as CaCO <sub>3</sub>	N/A	19.0, 21.8
Conductivity, $\mu$ mho/cm @ 25° C	N/A	79.7, 86.5
Total dissolved solids, mg/L	--	--
Calcium, mg/L Ca	N/A	2.89, 2.78
Hardness, mg/L as CaCO <sub>3</sub>	N/A	7.23*, 6.95*
Temperature, °C	N/A	10.3, 12.8
Chloride, mg/L	--	--
Sulfate, mg/L	--	--

\*Calculated

## 5. Distribution System:

Does the distribution system contain lead service lines?

☐ yes ☒ no

If your system has lead service lines, mark below the approximate number of lines which can be located from existing records.

None \_\_\_\_\_ Some \_\_\_\_\_ Most \_\_\_\_\_ All \_\_\_\_\_

Is the distribution system flushed?

☐ None ☒ Some ☐ Most ☐ All



6. Historical Information:

Is there a history of water quality complaints?

☐ yes ☒ no

If yes, then answer the following:

Are the complaints documented? ☐ yes ☐ no

Mark the general category of complaints below. Use:

1 for some complaints in this category

2 for several complains in this category

3 for severe complaints in this category

Categories of complaints:

Taste and odor \_\_\_\_\_

Color \_\_\_\_\_

Sediment \_\_\_\_\_

Other (specify) \_\_\_\_\_

Have there been any corrosion control studies?

☐ yes ☒ no

If yes, please indicate:

Date(s) of study From \_\_\_\_\_ To \_\_\_\_\_

Study conducted by PWS personnel? ☐ yes ☐ no

Brief results of study were \_\_\_\_\_

Study results attached? ☐ yes ☐ no

Were treatment changes recommended? ☐ yes ☐ no

If yes:

Were treatment changes implemented? ☐ yes ☐ no

Have corrosion characteristics of the treated water changed? ☐ yes ☐ no

If yes, how has change been measured?

General observation \_\_\_\_\_

Coupons \_\_\_\_\_

Frequency of complaints \_\_\_\_\_

Other \_\_\_\_\_ Briefly indicate below:

\_\_\_\_\_

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## 7. Treatment Constraints:

Optimal corrosion control treatment means the corrosion control treatment that minimizes the lead and copper concentrations at users' taps while insuring that the treatment does not cause the water system to violate any State or national primary drinking water regulations. Please indicate below which constraints to treatment will apply to your PWS. Use the following code:

- 1 Some constraint = Potential Impact but Extent is Uncertain.
- 2 Significant constraint = Other Treatment Modifications Required to Operate Option.
- 3 Severe constraint = Additional Capital Improvements Required to Operate Option.
- 4 Very severe constraint = Renders Option Infeasible.

Constraint	Treatments			
	pH/Alkalinity adjustment	Calcium adjustment	Inhibitor PO <sub>4</sub>	Si
A. Regulatory	1	1	1	1
SOCs/IOCs	1	1	1	1
SWTR: Turbidity	1	1	1	1
Total Coliforms	1	1	1	1
SWTR/GWDR: Disinfection	1	1	1	1
Disinfection Byproducts	1	1	1	1
Lead and Copper Rule	1	1	1	1
Radionuclides	1	1	1	1
B. Functional	1	1	1	1
Taste & Odor	1	1	1	1
Wastewater Permit	1	1	1	1
Aesthetics	1	1	1	1
Operational	1	1	1	1
Other	1	1	1	1

## 8. Evaluation:

Briefly summarize the review of the corrosion control literature that pertains to your PWS. A report or summary can be appended to this form if preferred.

N/A

Were other similar facilities located which are experiencing successful corrosion control? N/A  
☐ yes ☐ no

If yes, identify their corrosion control treatment method.

None \_\_\_\_\_  
pH/Alkalinity adjustment \_\_\_\_\_  
Calcium adjustment \_\_\_\_\_  
Inhibitor \_\_\_\_\_  
    Phosphate based \_\_\_\_\_  
    Silica based \_\_\_\_\_

## 9. Recommendations:

The corrosion control treatment method installed or being proposed is:

pH/Alkalinity adjustment X  
Target pH is 8.5 units  
Target alkalinity is 80 mg/L as  $\text{CaCO}_3$

Calcium adjustment X  
Target calcium concentration is 40 mg/L Ca

Inhibitor N/A  
    Phosphate based \_\_\_\_\_  
        Brand name \_\_\_\_\_  
        Target dose \_\_\_\_\_ mg/L  
        Target residual \_\_\_\_\_ mg/L orthophosphate as P  
    Silica based \_\_\_\_\_  
        Brand name \_\_\_\_\_  
        Target dose \_\_\_\_\_ mg/L  
        Target residual \_\_\_\_\_ mg/L as  $\text{SiO}_2$

Rationale for the proposed corrosion control treatment is:\*

Discussed in the enclosed report

EPA Revised Guidance Manual for Selecting Lead & Copper Control Strategies

Briefly explained below

\*Non-compliance data is based upon 2007 lead and copper monitoring data which had not been issued a NOV by DENR until now (2011). This system has been in consistent compliance for the last four (4) years and should be eligible for reduced monitoring. It is suspected that inadequate flushing of the distribution lines may have been the cause of the 2007 non-compliance. It is recommended that the system apply for reduced monitoring, carefully monitored (i.e. pH, etc.) and no further action be taken until a lead and copper exceedance can be documented.



List your proposed or operating guidelines:

<u>Parameter</u>	<u>Operating Range</u>
<u>pH</u>	<u>7.0 – 8.5</u>
<u>Alkalinity</u>	<u>50 – 75 mg/L</u>
<u>Hardness</u>	<u>30 – 50 mg/L</u>
<u>      </u>	<u>      </u>
<u>      </u>	<u>      </u>
<u>      </u>	<u>      </u>
<u>      </u>	<u>      </u>

Briefly explain why these guidelines were selected.

To improve the buffering capacity and pH stabilization of source water should raw water quality deteriorate.

10. Please provide any additional comments that will assist in determining optimal corrosion control treatment for your PWS.

See explanation in section 9. Note that no further action is recommended other than applying for reduced monitoring.